

Buried Polar Residues in Coiled-Coil Interfaces<sup>†,‡</sup>

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**ABSTRACT:** Coiled coils, estimated to constitute 3–5% of the encoded residues in most genomes, are characterized by a heptad repeat,  $(abcdefg)_n$ , where the buried *a* and *d* positions form the interface between multiple  $\alpha$ -helices. Although generally hydrophobic, a substantial fraction ( $\sim 20\%$ ) of these *a*- and *d*-position residues are polar or charged. We constructed variants of the well-characterized coiled coil GCN4-p1 with a single polar residue (Asn, Gln, Ser, or Thr) at either an *a* or a *d* position. The stability and oligomeric specificity of each variant were measured, and crystal structures of coiled-coil trimers with threonine or serine at either an *a* or a *d* position were determined. The structures show how single polar residues in the interface affect not only local packing, but also overall coiled-coil geometry as seen by changes in the Crick supercoil parameters and core cavity volumes.

Coiled coils are an oligomerization motif commonly occurring at the interface between separate protein chains. Coiled coils are found in many cytoskeletal and contractile systems (e.g., intermediate filaments, nuclear lamins, and myosin), transcription regulators (e.g., Myc and Max, Fos and Jun, GCN4), viral envelope proteins (e.g., MoMLV, HIV, SIV, influenza), and other systems (1). Genome database searches with coiled-coil prediction programs suggest that 3–5% of all protein residues exist as coiled coils (2). This implies a role for the coiled-coil motif in a wide range of biological functions. A more complete understanding of the interactions significant to coiled-coil formation will be useful in analyzing protein–protein interactions, predicting candidate partners, and assigning potential biological functions to novel proteins.

Coiled coils are comprised of two or more  $\alpha$ -helices that wrap around each other. The hallmark of coiled coils is a heptad repeat of amino acids,  $(abcdefg)_n$ , with a predominance of hydrophobic residues at the buried first and fourth (*a* and *d*) positions and charged residues frequently at the fifth and seventh (*e* and *g*) positions (Figure 1). Despite the prevalence of hydrophobic residues, approximately 20% of the *a* and *d* residues are polar or charged (2–6).

Previous work shows that buried polar residues in coiled coils can be important determinants of structural uniqueness (7–12), influencing both oligomeric state (e.g., dimers vs trimers) and strand orientation (i.e., parallel vs antiparallel). For example, *a*-position asparagines and lysines, both more frequent in coiled-coil dimers than trimers (Table 1), have been found to specify dimeric coiled coils (8, 10). In addition,

it has been shown that buried asparagines can direct strand orientation (9).

To address the role of buried polar residues in coiled-coil assembly, we constructed and characterized a series of peptides based on the GCN4-p1 sequence (13, 14) (a well-studied model system for coiled coils) with a single polar residue at either the *a* or the *d* position (Figure 1). We characterized these peptides in terms of their helical content, thermal stability, and oligomeric specificity. To investigate how polar residues are accommodated in the interior of coiled coils, we determined the crystal structures of four coiled-coil trimers containing a serine or threonine at either the *a* or the *d* position.

## EXPERIMENTAL PROCEDURES

**Peptide Synthesis and Purification.** The sequence of the GCN4-pVL variants is Ac-RMKQLEDKVEE-L/X-LSK-V/X-YHLENEVARLK<sup>**d**</sup>KL<sup>**a**</sup>VGER, where the *a* and *d* positions are underlined. Position 12(*d*) (boldface) is either a leucine or a polar residue (N, Q, S, or T), and position 16(*a*) (boldface) is either a valine or a polar residue. Peptides were synthesized as described previously (15) using Fmoc chemistry, desalted over a Sephadex G-25 column in 5% acetic acid, and purified by reverse-phase HPLC using a C18 preparative column. Purity was checked by analytical HPLC with a 0.1%/min water/acetonitrile gradient with 0.1% trifluoroacetic acid, and the mass of the peptides was verified by MALDI-TOF<sup>1</sup> mass spectroscopy (PerSeptive Biosystems, Inc., Framingham, MA).

**Circular Dichroism (CD) and Sedimentation Equilibrium.** CD measurements were carried out in 150 mM NaCl, 50

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<sup>‡</sup> Coordinates have been deposited in the RCSB Protein Data Bank: pVL<sub>S</sub>, 1IJ0; pVL<sub>T</sub>, 1IJ1; pVL<sub>L</sub>, 1IJ2; pVL<sub>S</sub>, 1IJ3.

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<sup>1</sup> Abbreviations: MALDI-TOF, matrix-assisted laser desorption ionization time-of-flight; CD, circular dichroism; CSP and LCSP, Crick supercoil parameters and local Crick supercoil parameters, respectively; *T*<sub>m</sub>, midpoint of thermal unfolding transition; RMSD, root-mean-square deviation.

mM Na<sub>2</sub>HPO<sub>4</sub>, pH 7.0, at 4 °C with an AVIV 62A or AVIV 62A/DS CD spectrometer (AVIV Instruments, Lakewood, NJ) using a 10 mm cell. Peptide concentrations were determined from the tyrosine absorption maxima at 276 nm in 6 M guanidine hydrochloride (16). Peptide concentrations were 10 μM for CD experiments. Thermal melts were performed using the same buffer conditions and peptide concentrations. Data for the thermal melts were acquired by averaging the CD signal over 30 s with a 90 s equilibration time after each 2 °C step. The midpoint of the thermal unfolding transition ( $T_m$ ) reported is the maximum of the first derivative of the thermal melt data.

Sedimentation equilibrium experiments were performed in the same buffer conditions as for CD measurements using a Beckman XLA-90 analytical ultracentrifuge (Beckman Coulter, Palo Alto, CA). Sedimentation equilibrium experiments were performed at peptide concentrations of 20, 50, and 100 μM. Data were collected after spinning for 18–24 h at speeds ranging from 18 000 to 35 000 rpm at 4 °C. For all speeds and time points, a second data set was acquired 3 h after the first. The two data sets were compared to ensure that the samples were at equilibrium. A single ideal species model was fit to the data. The partial specific volume was calculated from the residue-weighted average of the amino acid sequence (17). Solvent density was calculated from the solvent composition (17).

**Crystallization and Data Collection.** Crystallization conditions were found using a sparse matrix screen (Hampton Research, Laguna Niguel, CA). The appropriate conditions were further optimized to yield crystals large enough for diffraction studies. The final crystallization conditions were as follows: pVL<sub>T</sub>, 0.1 M sodium cacodylate, pH 6.0, 0.2 M zinc acetate, 20% PEG 8000; pVL<sub>S</sub>, 0.1 M sodium cacodylate, pH 7.0, 0.2 M zinc acetate, 16% PEG 8000; pV<sub>T</sub>L, 0.1 M HEPES, pH 7.0, 25 mM cadmium sulfate, 1 M sodium acetate; pV<sub>S</sub>L, 0.1 M MES, pH 6.5, 50 mM cesium chloride, 10% PEG 20,000. Three of the crystals were soaked for a brief period in stabilizing solutions containing mother liquor and added cryoprotectant (pVL<sub>S</sub>, 40% MPD; pV<sub>T</sub>L, 20% MPD; pV<sub>S</sub>L, 40% MPD) before freezing. Crystals were flash-frozen and data collected at 100 K using either an in-house source (RU-300 rotating anode with an R-AXIS IV detector), beam-line X4A at the National Synchrotron Light Source (NSLS), Brookhaven, NY, with an R-AXIS IV detector, or beam-line 5.0.2 at the Advance Light Source (ALS), Lawrence National Laboratory, Berkeley, CA, with an ADSC-4 CCD detector (Table 2). Diffraction intensities were integrated using the programs DENZO and SCALEPACK (18) and reduced to structure factors with the program TRUNCATE from the CCP4 program suite (19).

**Model Building and Refinement.** Molecular replacement solutions were found using the program AMORE (19) with either the GCN4-p1 dimer (14) or the GCN4-pV<sub>Q</sub>L trimer (20) structure as models. For molecular replacement, the model structures were truncated to alanine at position 16(*a*) for all models and also at position 12(*d*) for pVL<sub>S</sub> and pVL<sub>T</sub>. In all cases, AMORE identified solutions using the GCN4-pV<sub>Q</sub>L trimer with significantly higher correlation and lower *R*-factors than those found using the GCN4-p1 dimer. This confirmed that asymmetric units of the crystals contained a coiled-coil trimer.

For GCN4-pVL<sub>S</sub>, the AMORE solution underwent simulated annealing using the program CNS (21) and then modified to a polyserine model, with the exception of the core residue, Ser 12(*d*), which was modified to alanine. The polyserine model was refined and water added using the program REFMAC (19). Water molecules were removed and side chains manually added to the model using program O (22) as the densities for the side-chain rotamers became apparent. The model was further refined with multiple rounds of manual improvement using O, followed by positional and *B*-factor refinement using CNS.

After finding the molecular replacement solution using AMORE, the pVL<sub>T</sub>, pV<sub>T</sub>L, and pV<sub>S</sub>L structures were refined by 10 independent rounds of simulated annealing. A  $2F_o - F_c$  map was computed in CNS using the structures from all 10 rounds. Density interpretation and manual improvement were done with the program O. Crystallographic refinement was done using the CNS program. The final structures were checked using simulated annealing omit maps (23). Density was not always apparent for residues 31–33 for some chains. When the density for these terminal residues was not observed, they were omitted from the model. In addition, the density for some side chains was not observed, and the occupancy on these side-chain atoms was reduced to zero.

## RESULTS

**Solution Studies.** The peptide GCN4-pVL was used as the reference peptide for our studies. GCN4-pVL is equivalent to GCN4-p1 with the *a*-position asparagine at residue 16 [denoted Asn 16(*a*)] mutated to valine. GCN4-pVL has valines at the *a* positions (except for a single methionine) and leucines at the *d* positions (Figure 1). The single methionine, Met 2(*a*), is present in wild-type GCN4-p1 and was not changed in these studies. As shown previously (7), GCN4-pVL is stable [ $T_m = 95$  °C at 10 μM as determined by circular dichroism (CD)] and forms a mixture of dimers and trimers as determined by sedimentation equilibrium analysis (Table 1). Eight variants of GCN4-pVL with four different polar residues (Asn, Gln, Ser, or Thr) at either an *a* (residue 16) or a *d* (residue 12) position were synthesized (Figure 1). The resulting peptides are termed pV<sub>X</sub>L for *a*-position variants and pVL<sub>X</sub> for *d*-position variants, where X is N, Q, S, or T, denoting the polar substitution. Wild-type GCN4-p1, equivalent to pV<sub>N</sub>L, is not renamed here.

Sedimentation equilibrium analyses of the peptides reveal only two variants with a unique oligomeric state (Table 1). pVL<sub>T</sub> is a trimer, and as noted previously (13), wild-type GCN4-p1 is a dimer. These results correlate with the residue distributions observed in coiled-coil databases: asparagines at the *a* position are more often found in dimeric coiled coils, while threonines at the *d* position are more frequent in trimeric coiled coils. Other buried polar residues do not exhibit clear oligomeric preferences as judged by sedimentation equilibrium, and the slight preferences observed do not correlate with database residue distributions. Polar substitutions in the *d* position result in a greater variation in  $T_m$  and helical content than substitutions in the *a* position (Table 1). While pV<sub>L</sub>N and pV<sub>L</sub>Q seem to favor dimer formation, the low helical content and low  $T_m$  of these peptides suggest that there may be a significant portion of predominantly unfolded monomer in solution, skewing the data toward

Table 1: Circular Dichroism and Sedimentation Equilibrium Data for GCN4-pVL and Polar Variants

peptide	polar substitution	$-\theta]_{222}$ ( $\times 10^3$ deg $\text{cm}^2$ $\text{dmol}^{-1}$ )	$T_m$ ( $^{\circ}\text{C}$ )	$\text{MW}_{\text{obs}}/\text{MW}_{\text{monomer}}^a$			database distributions <sup>b</sup>	
				20 $\mu\text{M}$	50 $\mu\text{M}$	100 $\mu\text{M}$	dimer (%)	trimer (%)
pVL	none	30.0	94	2.5	2.6	2.8	—	—
pV <sub>N</sub> L <sup>c</sup>	Asn 16(a)	30.9	54	2.0	2.1	2.1	3.9	0.7
pV <sub>T</sub> L	Thr 16(a)	30.3	68	2.4	2.4	2.6	0.7	5.6
pV <sub>S</sub> L	Ser 16(a)	28.5	52	2.3	2.3	2.4	2.0	5.1
pV <sub>Q</sub> L	Gln 16(a)	30.7	56	2.3	2.3	2.4	0.9	2.4
pVL <sub>N</sub>	Asn 12(d)	15.2	22	2.0	2.2	2.3	1.3	3.1
pVL <sub>T</sub>	Thr 12(d)	30.5	62	3.0	3.1	3.2	3.6	8.2
pVL <sub>S</sub>	Ser 12(d)	32.0	44	2.6	2.6	2.8	4.1	2.7
pVL <sub>Q</sub>	Gln 12(d)	25.8	26	2.1	2.1	2.2	1.8	1.6

<sup>a</sup>  $\text{MW}_{\text{obs}}/\text{MW}_{\text{monomer}}$  is the molecular weight determined from sedimentation equilibrium analysis divided by the expected mass of a monomer.

<sup>b</sup> Database distributions are the frequencies of the specified polar residue at a given heptad position in dimeric and trimeric coiled-coil databases, respectively (2). <sup>c</sup> pV<sub>N</sub>L is equivalent to GCN4-p1.

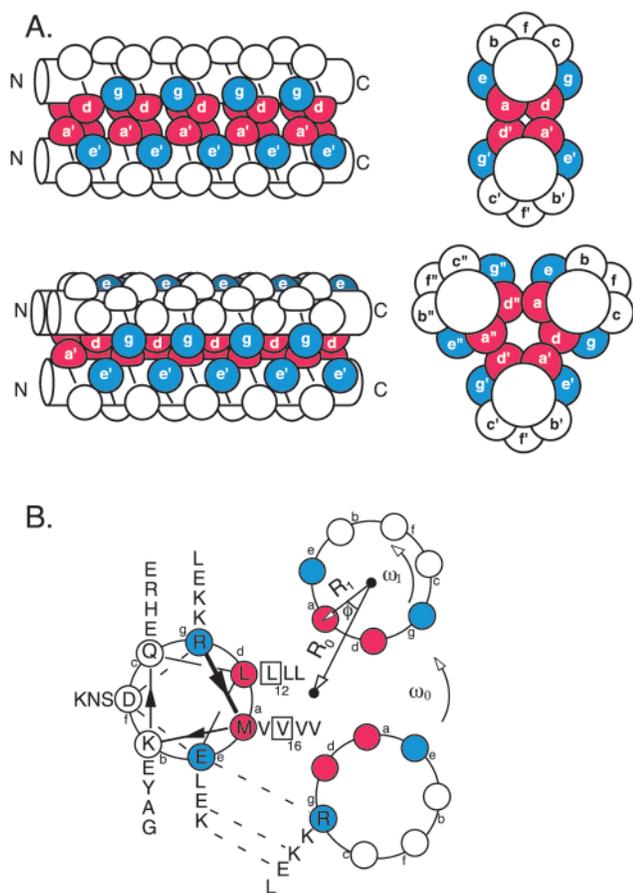


FIGURE 1: (A) Schematic view of dimeric (top) and trimeric (bottom) coiled coils. The *a* and *d* core positions are indicated in red and the *e* and *g* positions in blue. (B) Coiled-coil schematic of GCN4-pVL peptide as a trimer. Residues 16(*a*) and 12(*d*), the positions of the polar substitutions, are boxed. Interhelical *g* to *e'* salt bridges, indicated with dashed lines, are from the GCN4-p1QI structure (28). The Crick supercoil parameters (CSP) are indicated as:  $R_0$  and  $R_1$ , the radius for the supercoil and  $\alpha$ -helix, respectively;  $\omega_0$  and  $\omega_1$ , the supercoil pitch and  $\alpha$ -helix pitch, respectively;  $\phi$ , the  $C_\alpha$  phase angle, defined as the angle between vectors from the  $\alpha$ -helix center to the supercoil center and to the  $C_\alpha$  of the *a*-position residue.

lower molecular weight species. Thus, it is likely that the molecular weight of the folded form of these peptides is underestimated by sedimentation equilibrium analysis. In general, our data are in agreement with the oligomeric preferences seen in other model coiled-coil systems (11, 12).

The data for pV<sub>Q</sub>L, which is found here and elsewhere (10) to exist as a mixture of dimers and trimers, are in conflict with other studies which suggest that *a*-position glutamines are sufficient to specify trimeric coiled coils (11). These differences may illustrate the importance of context when accounting for the effects of buried polar residues on structural uniqueness.

**Crystal Structures.** The structures of coiled-coil trimers with threonines and serines at either the *a* or the *d* positions (pV<sub>T</sub>L, pV<sub>S</sub>L, pVL<sub>T</sub>, pVL<sub>S</sub>) were determined at a resolution of 1.9 Å or higher (Table 2). Although three of these four peptides exist as mixed oligomeric species in solution, all four peptides crystallized as a trimer. All four structures show an approximate, noncrystallographic, 3-fold symmetry. As noted, pVL<sub>T</sub> has unique trimeric specificity and is the most stable *d*-position polar variant (Table 1). The structure of pVL<sub>T</sub> (Figures 2A and 3) shows symmetric packing of the Thr 12(*d*) residues with an ordered hydrogen bond network in the core. The hydroxyl group of each Thr 12(*d*) makes an identical *i* to *i* - 4 backbone hydrogen bond to the peptide carbonyl of Lys8(*g*), directing the threonine methyl groups into the core (Figure 3). Four structured water molecules, between the Val 9(*a*) and Thr 12(*d*) layers, form the vertexes of a trigonal pyramid. The three outer water molecules form potential hydrogen bonds to the Val 9(*a*) backbone carbonyl. As there are more potential hydrogen bonds than hydrogen atoms in this network (12 vs 11), partial water molecule occupancies and transient hydrogen bonds are expected.

Similar to the *d*-position threonine residues, *d*-position serines in pVL<sub>S</sub> are arranged symmetrically. In contrast to the *d*-position threonines, the hydroxyl groups of the core serines are directed toward the center to form an internal hydrogen bond network (Figures 2B and 3). No structured water molecules are observed in the core structure. In the vicinity of the serines there is a local decrease in the superhelical radius of about 0.5 Å (Figure 4A), changes in the supercoil pitch (Figure 4B), and an asymmetric rotamer distribution at Val 9(*a*) (Figure 3).

In contrast to the *d*-position polar residues, the *a*-position serines and threonines are not packed with 3-fold symmetry and are in different rotamers. The serines in pV<sub>S</sub>L have the most structural variability, with one hydroxyl group directed toward the exterior, another making an *i* to *i* - 3 hydrogen bond to the peptide carbonyl of Leu 13(*e*), and the third with the partial occupancy of two alternate rotamers, either directed to the core or pointing outward toward the bulk

Table 2: Data Collection, Refinement Statistics, and Crick Supercoil Parameters

crystal	pVL <sub>T</sub> Thr 12( <b>d</b> )	pVL <sub>S</sub> Ser 12( <b>d</b> )	pV <sub>T</sub> L Thr 16( <b>a</b> )	pV <sub>S</sub> L Ser 16( <b>a</b> )
space group	<i>P</i> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub>	<i>P</i> 2 <sub>1</sub> 2 <sub>1</sub>	<i>P</i> 1
<i>a</i> , <i>b</i> , <i>c</i> (Å)	25.7, 42.2, 42.7	35.5, 28.6, 44.5	35.8, 46.3, 51.4	25.6, 25.8, 34.4
$\alpha$ , $\beta$ , $\gamma$ (deg)	90.0, 98.9, 90.0	90.0, 109.6, 90.0	90.0, 90.0, 90.0	86.5, 84.5, 79.4
data collection				
X-ray source	in-house	in-house	ALS	NSLS
resolution (Å)	1.86	1.86	1.70	1.80
total observations	80296	70770	63455	23291
unique reflections	7316	7255	9768	7378
completeness (%)	95.3 (94.4) <sup>f</sup>	94.4 (93.7) <sup>f</sup>	99.3 (99.2) <sup>f</sup>	92.6 (89.0) <sup>f</sup>
<i>R</i> <sub>merge</sub> <sup>a</sup>	0.046 (0.306) <sup>f</sup>	0.065 (0.315) <sup>f</sup>	0.048 (0.283) <sup>f</sup>	0.059 (0.180) <sup>f</sup>
refinement				
<i>R</i> <sub>cryst</sub> <sup>b</sup>	0.227	0.211	0.243	0.220
<i>R</i> <sub>free</sub> <sup>c</sup>	0.278	0.258	0.279	0.269
RMSD bonds (Å)	0.010	0.014	0.014	0.009
RMSD angles (deg)	1.3	1.9	1.6	1.3
Crick supercoil parameters (CSP) <sup>d</sup>				
$\alpha$ carbon phase, $\phi$ (deg)	16.1	18.2	18.9	19.7
supercoil radius, <i>R</i> <sub>0</sub> (Å)	6.25	6.03	6.25	6.29
supercoil pitch (residues/turn)	101	88	98	99
supercoil deviation, RMSD (Å) <sup>e</sup>	0.292	0.408	0.256	0.265

<sup>a</sup>  $R_{\text{merge}} = \sum_j |I_j(hkl) - \langle I(hkl) \rangle| / \sum_j \langle I(hkl) \rangle$ , where  $I_j$  is the intensity measurement for reflection ( $hkl$ ) and  $\langle I \rangle$  is the mean intensity over  $j$  measurements. <sup>b</sup>  $R_{\text{cryst}} (R_{\text{free}}) = \sum_j |F_{\text{obs}}(hkl) - F_{\text{calc}}(hkl)| / \sum_j F_{\text{obs}}(hkl)$ , where  $F_{\text{obs}}$  and  $F_{\text{calc}}$  are observed and calculated structure factors, respectively. No  $\sigma$ -cutoff was applied. <sup>c</sup> Ten percent of the reflections were excluded from refinement and used to calculate  $R_{\text{free}}$ . <sup>d</sup> The Crick supercoil parameters (CSP) were determined by fitting an ideal coiled coil to the central three heptads of the final structure using a user-defined CHARMM function (see Results) (24, 25). <sup>e</sup> Supercoil RMSD is between the  $C_{\alpha}$  atoms of the model coiled coil and the observed structure. <sup>f</sup> Values in parentheses correspond to the highest resolution shell: 1.94–1.86 Å for pVL<sub>T</sub> and pVL<sub>S</sub>, 1.76–1.70 Å for pV<sub>T</sub>L, and 1.86–1.80 Å for pV<sub>S</sub>L.

solvent (Figure 2D). Two coordinated water molecules are observed in the core. The hydroxyl groups of threonine in pV<sub>T</sub>L make intrahelical peptide carbonyl contacts, but with some variation: all three make an  $i - 3$  intrahelical contact, but one also makes the alternative  $i - 4$  contact observed for **d**-position threonines (Figure 2C).

**Supercoil Parameters.** The geometry of a regular coiled coil is defined by a simple set of parameters [denoted Crick supercoil parameters (CSP)] that include the radius ( $R_0$  and  $R_1$ ) and pitch ( $\omega_0$  and  $\omega_1$ ) for both the supercoil and  $\alpha$ -helix, respectively, and the **a**-position  $C_{\alpha}$  phase angle ( $\phi$ ) (Figure 1) (24, 25). To examine how single-residue substitutions affect the coiled-coil geometry, we determined the CSP values of each coiled coil (Table 2) using the central three heptads (residues 5–25) (24, 25). We fit only the central three heptads because the regions outside these residues are less structured and fit poorly to ideal supercoil parameters (Figure 4D). For each set of superhelical parameters, we determined the  $C_{\alpha}$  root-mean-square deviation (RMSD) between the actual structure and a model coiled coil built using the CSP values. Three of the structures fit closely to the model coiled coil, with RMSDs less than 0.29 Å. pVL<sub>S</sub>, however, deviates more substantially, with an RMSD of 0.41 Å (Table 2).

To investigate local effects of the buried polar residues, we utilized a heptad-specific, local Crick supercoil parameter (LCSP), determined by fitting only seven residues at a time. This local fitting of coiled-coil parameters—sliding the window in one-residue increments—allows detection of changes in structure not readily apparent by visual inspection or global fitting (Figure 4). Our method extends previously described local fitting methods (26) by simultaneously determining multiple parameters for the coiled coil. Polar residues in the **a** position seem to have a smaller effect on the LCSP than **d**-position polar residues, as judged by the

observation that LCSP values for the **a**-position variants follow a common trend, while **d**-position LCSP values differ more widely (Figure 4). As noted earlier, **d**-position polar variants also vary more in stability than **a**-position polar variants. Local structural variations include a decrease in  $R_0$  centered around the buried serine residues in pVL<sub>S</sub>, and variations in  $\phi$  for both pVL<sub>S</sub> and pVL<sub>T</sub>. The observation that **a**-position polar residues have a smaller effect on supercoil parameters is also supported by comparisons of GCN4-pV<sub>Q</sub>L (10) with the structures shown here, and between the GCN4-pII (27) and GCN4-pIQI (28) trimer structures (data not shown).

**Cavity Volumes in the Core.** An analysis of coiled-coil trimers reveals sizable cavities in the cores (Figure 5). The cavities vary in volume, position, and structured water molecules, particularly around the buried polar positions. Generally, cavities are found in trimeric coiled coils between the **a** and **d** layers, although pV<sub>S</sub>L also has a cavity that extends from the position 12(**d**) layer to the position 19(**d**) layer (Figure 5D). There are no cavities observed between the position 12(**d**) layer and the position 16(**a**) layer for either pVL<sub>T</sub> or pVL<sub>S</sub> (Figure 5A,B).

One might expect **d**-position serine variants to have larger cavity volumes than the threonine variants. However, due to differences in supercoil radius and pitch (Table 2 and Figure 4), the cavities for pVL<sub>S</sub> are in fact smaller than the corresponding cavities in pVL<sub>T</sub> (Figure 5A,B). The largest cavity in pVL<sub>T</sub> contains structured water molecules. The **a**-position polar variants vary more in cavity size and position. Although pV<sub>S</sub>L has two fewer cavities than pV<sub>T</sub>L, the extended cavity in pV<sub>S</sub>L accounts for space that contains two cavities in pV<sub>T</sub>L (Figure 5C,D). This extended cavity is the largest in all four structures and contains two structured water molecules. The cavities of pV<sub>T</sub>L are the smallest and most regular.

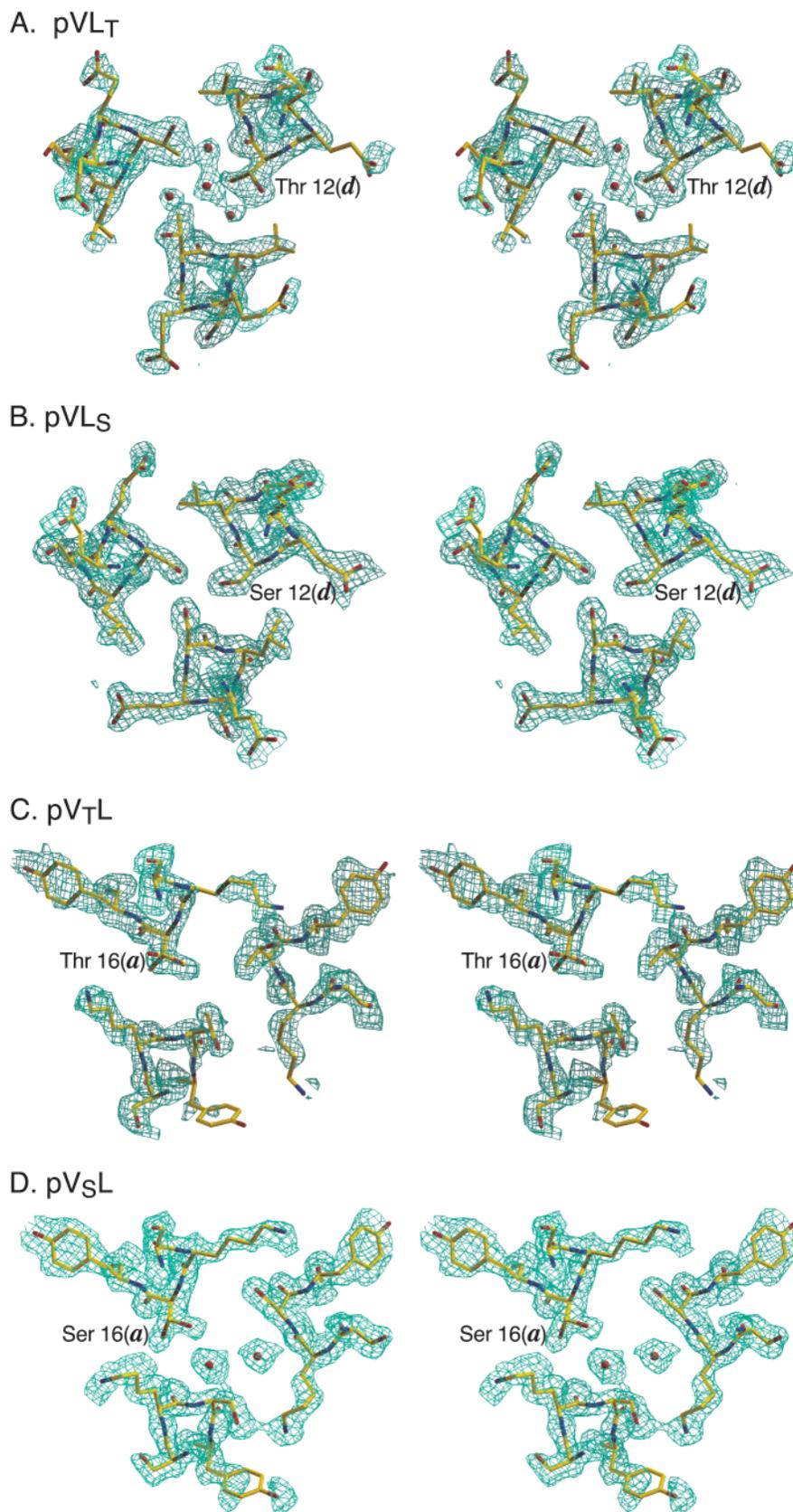


FIGURE 2: Stereodiagrams of selected area composite omit electron density maps superimposed onto the final refined structures. Maps are contoured at  $1\sigma$ . Water molecules are represented as red spheres. (A) pVL<sub>T</sub>; (B) pVL<sub>S</sub>; (C) pV<sub>T</sub>L, with alternate rotamers for the labeled Thr 16(*a*); and (D) pV<sub>S</sub>L, with alternate rotamers for the labeled Ser 16(*a*).

*Interhelical Surface Contacts.* Ionic surface interactions between helices are important in both dimeric and trimeric coiled coils and are determinants of coiled-coil specificity

(29, 30). Nevertheless, it is commonly observed that not all possible interhelical interactions are formed. The four structures presented here all have identical surface residues,

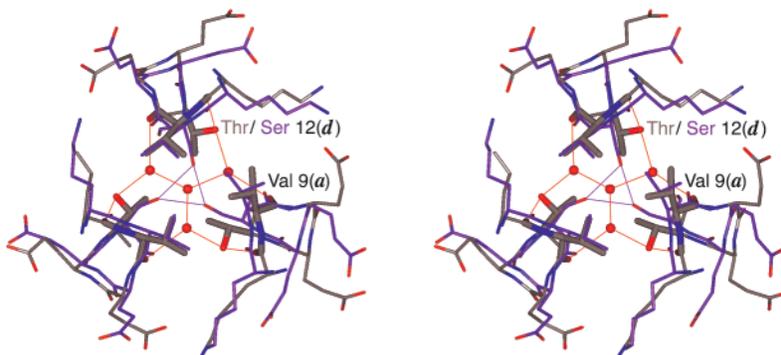


FIGURE 3: Stereodiagram of pVL<sub>S</sub> (residues 8–12, purple backbone) superimposed on pVL<sub>T</sub> (gray backbone) including the polar *d* position (residue 12). Val 9(*a*) and Thr 12(*d*) of pVL<sub>T</sub> are shown as thicker rods. A hydrogen bond network, shown in red, is observed between water molecules (red spheres), all Thr 12(*d*) residues, and the peptide carbonyl oxygens of residues 8 and 9 in the pVL<sub>T</sub> structure. The decreased supercoil radius of pVL<sub>S</sub> is apparent. The labeled Val 9(*a*) of pVL<sub>S</sub> adopts a unique rotamer not seen for any other *a*-position valines.

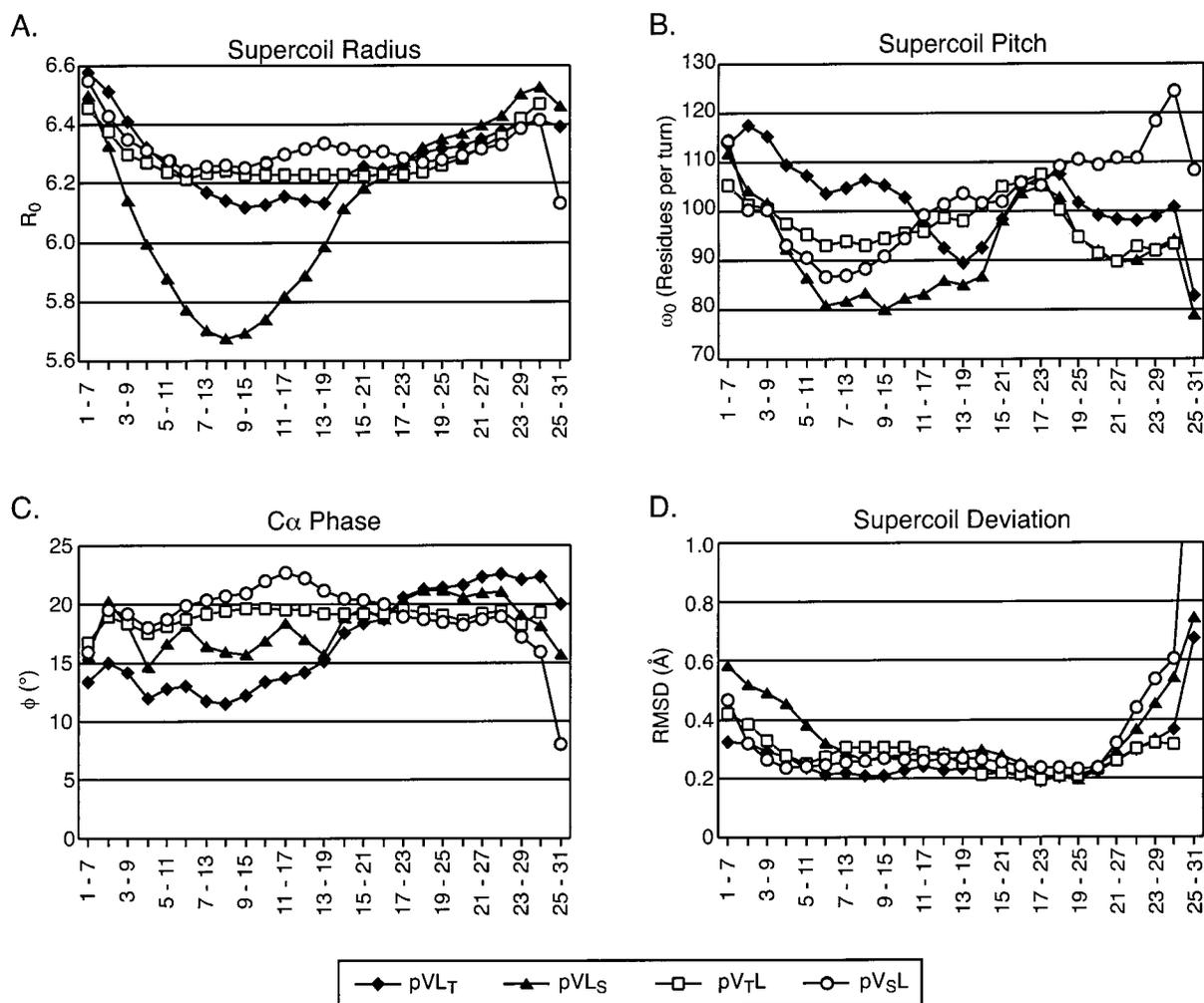


FIGURE 4: Local Crick supercoil parameters (LCSP), as defined in Figure 1, plotted for pVL<sub>T</sub> (closed diamonds), pVL<sub>S</sub> (closed triangles), pV<sub>T</sub>L (open squares), and pV<sub>S</sub>L (open circles). The indicated heptads were used to determine LCSP values. (A) Supercoil radius ( $R_0$ ) in angstroms; (B) supercoil pitch ( $\omega_0$ ) in residues per turn; (C)  $\phi$  in degrees; and (D) deviation from ideal supercoil.

but different surface contact patterns (Table 3). Some seemingly repulsive charge–charge interactions are allowed because of the presence of a third, mutually attractive, charged group forming a salt bridge triad (e.g., the A-chain Arg 1 to B-chain Lys-3 interaction in pVL<sub>S</sub> is mediated by the B-chain Glu-6 carboxylate).

While the variation in surface residue contacts may be a result of crystal packing, all four trimers required different

crystallization conditions and crystallized in different space groups (Table 2), suggesting that differences between the four trimers exist in solution. Differences in supercoil radius and pitch may partially account for variations in surface interhelical interactions, as subtle shifts in these parameters may be enough to bring potentially interacting residues (e.g., *e* and *g* positions) closer together or further apart. These differences illustrate that *ab initio* design and structure

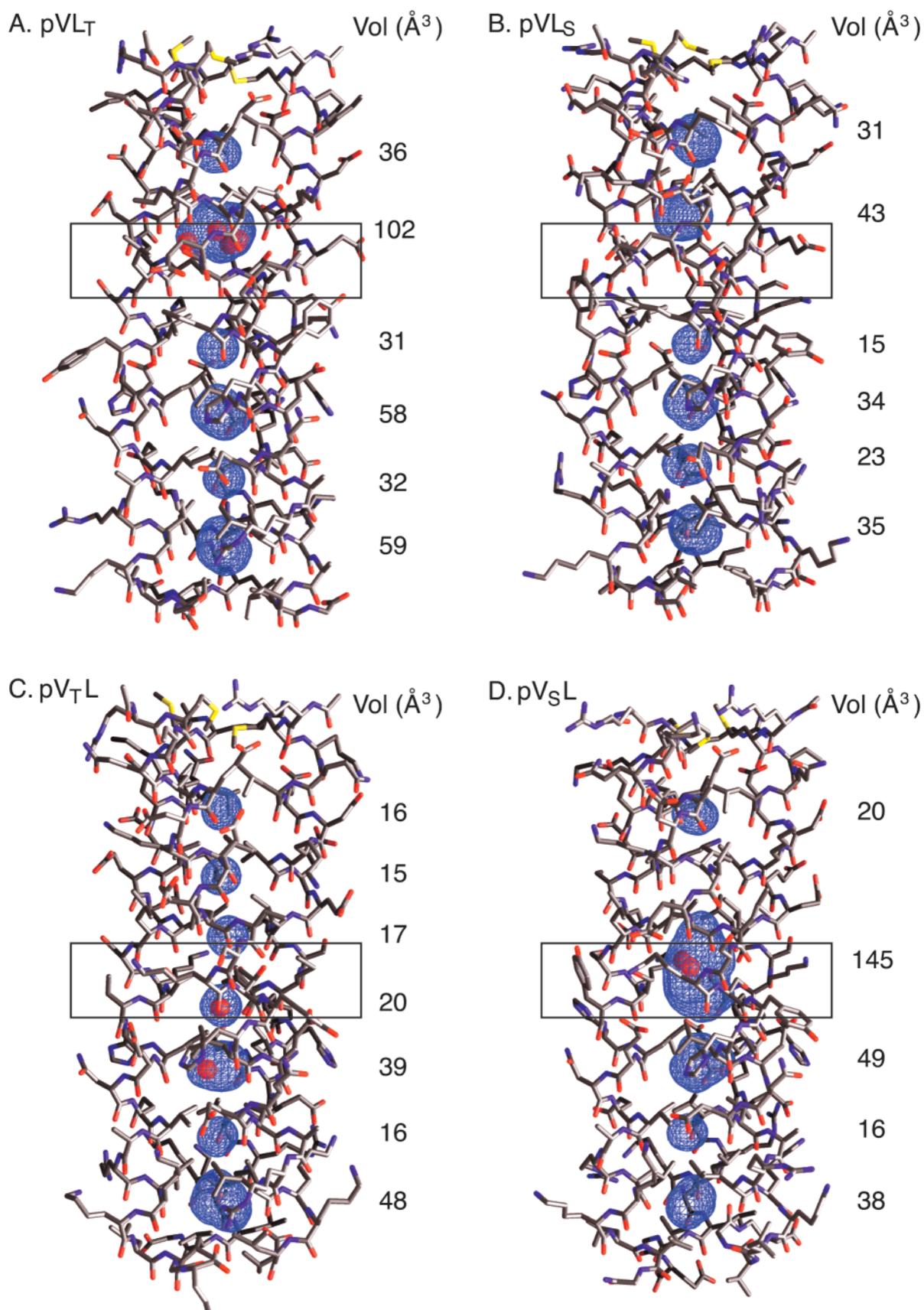


FIGURE 5: Cavities in the cores of coiled-coil trimers: (A) pVL<sub>T</sub>, (B) pVL<sub>S</sub>, (C) pV<sub>T</sub>L, and (D) pV<sub>S</sub>L. Vol, cavity volume ( $\text{\AA}^3$ ). Cavities were from molecular surface calculated with the program GRASP (36). Water molecules were omitted for these calculations. The cross-sections containing the core polar residues (see also Figure 2) are indicated with boxes. There are four water molecules (red spheres) in the cavity of pVL<sub>T</sub> (A), two in pV<sub>T</sub>L (C), and two in pV<sub>S</sub>L (D).

Table 3: Interhelical Interactions between Charged and Polar Groups<sup>a</sup>

interhelical contacts		interaction distances (Å)			
type	residues	pVL <sub>T</sub>	pVL <sub>S</sub>	pV <sub>T</sub> L	pV <sub>S</sub> L
<i>g-e'</i>	Arg A1(N <sub>η1</sub> )–Glu B6(O <sub>ε1</sub> )	2.8	3.6	3.0	
<i>g-e'</i>	Arg B1(N <sub>η2</sub> )–Glu C6(O <sub>ε1</sub> )	2.7	3.0	3.3	2.7
<i>g-e'</i>	Arg C1(N <sub>η2</sub> )–Glu A6(O <sub>ε2</sub> )	W		3.1	2.7
<i>g-e'</i>	Lys A15(N <sub>ζ</sub> )–Glu B20(O <sub>ε1</sub> )	2.6			3.7
<i>g-e'</i>	Lys C15–Glu A20	W	W		W
<i>g-e'</i>	Glu A22(O <sub>ε1</sub> )–Lys B27(N <sub>ζ</sub> )	2.9	2.9	3.6	2.7
<i>g-e'</i>	Glu B22(O <sub>ε2</sub> )–Lys C27(N <sub>ζ</sub> )	2.7	3.0	2.7	3.5
<i>g-e'</i>	Glu C22(O <sub>ε1</sub> )–Lys A27(N <sub>ζ</sub> )	2.7	2.7	2.8	
<i>g-b'</i>	Arg A1(N <sub>η2</sub> )–Lys B3(N <sub>ζ</sub> )		3.0		
<i>g-b'</i>	Lys A8(N <sub>ζ</sub> )–Glu B10(O <sub>ε2</sub> )	2.9		2.9	
<i>g-b'</i>	Lys B8(N <sub>ζ</sub> )–Glu C10(O <sub>ε2</sub> )	2.5	W		
<i>g-b'</i>	Lys C8(N <sub>ζ</sub> )–Glu A10(O <sub>ε2</sub> )	3.7			
<i>g-a'</i>	Lys A15(N <sub>ζ</sub> )–Ser B16(O <sub>γ</sub> )				3.4
<i>g-a'</i>	Lys C15(N <sub>ζ</sub> )–Thr A16(O <sub>γ1</sub> )			2.8	
<i>e-g'</i>	Glu A6(O <sub>ε1</sub> )–Lys C8(N <sub>ζ</sub> )		3.8		
<i>c-e'</i>	Gln A4(N <sub>ε2</sub> )–Glu B6(O <sub>ε2</sub> )		3.9	W	
<i>c-e'</i>	Arg A25(N <sub>η2</sub> )–Lys B27(N <sub>ζ</sub> )	2.8	W		
<i>c-e'</i>	Arg B25(N <sub>η2</sub> )–Lys C27(N <sub>ζ</sub> )				2.6
<i>c-b'</i>	Glu A11(O <sub>ε2</sub> )–Tyr B17(O <sub>η</sub> )		2.8		

intrahelical contacts		interaction distances (Å)			
type	residues	pVL <sub>T</sub>	pVL <sub>S</sub>	pV <sub>T</sub> L	pV <sub>S</sub> L
<i>b-e</i>	Lys B3(N <sub>ζ</sub> )–Glu B6(O <sub>ε2</sub> )		3.0		
<i>g-c</i>	Glu A22(O <sub>ε2</sub> )–Arg A25(N <sub>η2</sub> )	2.9	W		

<sup>a</sup> Interactions listed as *g-e'* indicate an interaction between the *g*-position residue of one chain and the *e*-position residue of an adjacent chain. Interacting residues are identified by amino acid type, chain designation (A, B, and C for the three chains), and residue number. Interaction distances are shown. A “W” indicates that a potential salt bridge interaction is mediated through a structured water molecule, but that the residues are too far apart to directly interact. The intrahelical contacts listed mediate an otherwise repulsive interhelical contact.

prediction of coiled coils may need to account for alternative ionic interactions among the surface residues.

## DISCUSSION

There are several possible reasons for the presence of buried polar residues in natural coiled coils. Buried polar residues may help specify the oligomeric state in sequences that would otherwise exist as a mixture of dimers, trimers, or higher order structures (6, 8, 10, 20). We find that determining oligomeric specificity may be a significant role of only a few buried polar residues, as only an *a*-position asparagine and a *d*-position threonine are found to specify unique oligomeric states. In contrast, other polar residues do not specify a unique oligomeric preference. As hydrophobic core sequences are an important determinant of oligomeric state (7), the effects of polar residues on oligomeric specificity may be limited and depend on the identities of the other core residues.

Buried polar residues in coiled coils likely serve important roles in determining structural uniqueness. Interactions between polar atoms may compensate for the unfavorable desolvation energy of buried polar residues. These interactions are commonly observed in protein cores and have been proposed to be an important source of structural specificity (8, 31). In the case of coiled coils, the selection of polar–polar pairings could ensure that composite  $\alpha$ -helices are aligned correctly in homotypic coiled coils (8, 14). In an analogous manner, polar–polar pairings may be a

discriminating factor in partner selection in heterotypic coiled coils. A preference for polar–polar pairing has been confirmed for *a*-position asparagines in a number of genetic selection experiments (32–34). In the structures presented here, the core-position threonines and serines satisfy their hydrogen bond requirements either with residues on adjacent strands (often mediated through water molecules), with the peptide backbone, with external residues and solvent, or combinations of the above. This ability to satisfy hydrogen bond requirements through several means may allow these polar residues flexibility in forming heterotypic coiled coils as they may pair with both polar and hydrophobic residues. The potential dual nature of polar residues suggests an avenue of further research in determining factors involved in heterotypic coiled-coil assembly.

Finally, a comparison of the four closely related peptide structures presented here illustrates the plasticity of coiled coils in response to polar substitutions. These structures show two distinct mechanisms for accommodating polar residues in the coiled-coil core. The first method is the filling of cavities with structured water molecules, and the second involves changes in the supercoil parameters, particularly the supercoil radius. Additionally, single polar residue core substitutions lead to variation in surface contacts. These variations may need to be accounted for in computational methods that have been successful at predicting and designing coiled coils, but that used cores that lacked polar residues (24, 35). In addition, these computational methods have assumed constant supercoil parameters along the length of the coiled coil. Our results show that superhelical parameters can be fit using local Crick supercoil parameters (LCSP), even as they vary along the length of the coiled coil. Utilization of such local fitting methods could play an important role in future coiled-coil design efforts.

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