Biodegradable Elastomers and Silicon Nanomembranes/Nanoribbons for Stretchable, Transient Electronics, and Biosensors

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Supporting Information

ABSTRACT: Transient electronics represents an emerging class of technology that exploits materials and/or device constructs that are capable of physically disappearing or disintegrating in a controlled manner at programmed rates or times. Inorganic semiconductor nanomaterials such as silicon nanomembranes/nanoribbons provide attractive choices for active elements in transistors, diodes and other essential components of overall systems that dissolve completely by hydrolysis in biofluids or groundwater. We describe here materials, mechanics, and design layouts to achieve this type of technology in stretchable configurations with biodegradable elastomers for substrate/encapsulation layers. Experimental and theoretical results illuminate the mechanical properties under large strain deformation. Circuit characterization of complementary metal-oxide-semiconductor inverters and individual transistors under various levels of applied loads validates the design strategies. Examples of biosensors demonstrate possibilities for stretchable, transient devices in biomedical applications.

KEYWORDS: Stretchable, flexible, transient, biodegradable electronics, biosensors

Advances in inorganic nanomaterials and unconventional fabrication techniques have recently yielded capabilities for broad classes of semiconductor devices and integrated systems that are entirely soluble in biofluids with nontoxic end products.1−17 Of particular interest are approaches that use single crystalline nanomembranes of silicon (Si NMs)8−10,14−16 or thin films of zinc oxide11 as the semiconductor to achieve not only the essential electronic components needed for integrated circuits but also other devices such as solar cells,8 temperature/strain/hydration sensors,8,15 multiplexed arrays of photodetectors,9 mechanical energy harvesters,11 and radio frequency power scavengers.9 For Si NMs, the rates of hydrolysis vary between tenths and tens of nanometers per day in groundwater, biofluids and other solutions of interest, where the specific rates depend on temperature, pH, ionic concentration, doping level of the silicon, and other factors.8,14,16 These and related electronic systems combine such semiconductors with Mg or other dissolvable metals (e.g., Zn, Fe, W, Mo) for interconnects and electrodes,18 MgO and

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SiO₂ or SiNₓ for dielectrics/encapsulating layers,⁸,¹⁹ biodegradable polymers (e.g., silk, poly lactic-co-glycolic acid (PLGA), polycaprolactone (PCL), polyactic acid (PLA)),⁸,¹⁵ or metal foils for substrates/packaging materials. In addition to their possibility for use in temporary biomedical implants, this type of physically “transient” device technology has potential applications in degradable environmental monitors/sensors, disposable consumer gadgets, “unrecoverable” electronics, and hardware-secure digital memories. Past work demonstrates capabilities for forming devices in rigid or flexible formats.⁴⁻¹¹,¹⁵ In many cases, particularly those that require integration with the human body, mechanical stretchability (i.e., ability to accommodate strains ≥1%) provides critically important advantages.²⁰,²¹ Here we describe transient electronic devices and sensors that use Si NMs and nanoribbons (NRs) in optimized configurations on a biodegradable elastomer (poly(1,8-octanediol-co-citrate), POC) to yield systems that can be reversibly stretched to strains of up to ∼30% with linear elastic mechanical responses. The electrical characteristics of the resulting devices are similar to those that use nontransient construction and rigid, planar substrates. Experimental and theoretical studies illustrate the underlying behaviors of the materials and their responses to mechanical deformation. Sensors of pH and skin-mounted monitors of electrophysiology (EP) provide two application examples.

Figure 1 presents optical micrographs and an exploded view schematic illustration of an array of stretchable, transient complementary metal-oxide-semiconductor (CMOS) inverters. Device fabrication begins with a series of doping processes on silicon on insulator (SOI, SOITEC, France) wafers, to create lightly doped p-well (p⁻) and heavily doped regions (p⁺, n⁺) for contacts of both p- and n-channel metal-oxide-semiconductor field-effect transistors (MOSFETs). Removal of the buried oxide with hydrofluoric acid (HF) releases the doped silicon device layer to yield isolated Si NMs. Transfer printing delivers these Si NMs onto a silicon wafer coated with a bilayer of poly(methyl methacrylate) (PMMA, Sigma-Aldrich, U.S.A.) and diluted polyimide (D-PI). Reactive ion etching (RIE, Plasmatherm) of the Si NMs with sulfur hexafluoride (SF₆) defines the active areas of the devices. Plasma-enhanced chemical vapor deposition (PECVD) of silicon dioxide (SiO₂) yields the gate and interlayer dielectrics. A patterned thin film of Mg formed by electron beam evaporation serves as source, drain, and gate electrodes as well as interconnects. A uniform overcoat of PECVD SiO₂ encapsulates the entire system (Figure 1, top left). Removing the PMMA by immersion in acetone enables release onto the surface of a
slab of polydimethylsiloxane (PDMS) (Figure 1, top right), for transfer onto a biodegradable elastomer substrate, POC (Figure 1, bottom right). Polydiol citrates represent a unique and interesting platform of elastomeric polyesters that exhibit versatility regarding fundamental and functional design strategies that can be used to meet the goals of a wide range of applications. Similar fabrication procedures are used for all of the various devices reported here. Details appear in Methods.

As explored in previous reports on nontransient stretchable Si NM electronics, layouts that include device islands connected by serpentine and/or noncoplanar interconnects allow the system to accommodate large deformations without fracture. Here, the interconnects undergo in- and out-of-plane buckling in response to tensile loads, whereas the islands remain flat and undeformed. The buckling process induces bending strains that are typically orders of magnitude smaller than the applied tensile strains. Additionally, the interconnects often involve metal encapsulated above and below by thin polymer layers, thereby placing the metal near the neutral mechanical plane, to minimize bending strains. We expect the fatigue properties under cyclic loads to be similar to those of related, but nontransient, stretchable systems. These same strategies prove valuable in the context of transient, stretchable Si NM systems introduced here. The results shown in Figure 1a illustrate arrays of stretchable, transient CMOS inverters with three different types of interconnects. Finite element analysis (FEA) reveals the mechanics associated with each case; a representative example appears in Figure 2a (see Supporting Information Figures S1–S3 and movies S1–S2 for additional information). For device islands that are bonded to the

Figure 2. Experimentally observed and calculated distributions of strain in stretchable, biodegradable CMOS inverters. Optical microscope images of CMOS inverters that consist of transistors joined by deformable interconnects for stretching up to ∼40% (top) and corresponding results from Finite Element Analysis (FEA, bottom). Cases for (a) no encapsulation, (b) soft encapsulation (0.1 MPa), and (c) hard encapsulation (2 MPa).
substrate and interconnects that are not, the elastic deformation is predicted to be >40%, determined from the distribution of strain in the Mg (bottom panels of Figure 2a). A top layer encapsulation of POC can enhance the bonding and protect the devices from the environment. The top panels in Figure 2b,c show optical images collected from experiments, where partially cured (Figure 2b, Young’s modulus of 0.1 MPa) and fully cured (Figure 2c, Young’s modulus of 2 MPa) layers of POC provide encapsulation. FEA shows that the interfacial stresses decrease as the Young’s modulus of encapsulation material increases, consistent with the role of this layer in preventing delamination. As shown in the bottom panels of Figure 2b,c, the maximum principal strain increases dramatically when freestanding interconnects are replaced by fully bonded ones. The maximum principal strain in the interconnects increases dramatically when freestanding interconnects are replaced by fully bonded ones. The maximum principal strain in the interconnects increases slightly from soft encapsulation to hard encapsulation. An ultrasoft encapsulation (Supporting Information Figure S2b, Young’s modulus of 0.1 KPa) leads to strains that differ by only <3% from the soft encapsulation case. This result highlights the value of substrate chemistries that allow strong, selective bonding only at the islands, as is well established for the case of a nontransient elastomer like silicone. Modifications to POC for this purpose represent topics of current work.

Figure 3. Evaluation of the distributions of strain in CMOS inverters interconnected in a filamentary serpentine (FS) mesh. (a) Schematic exploded view illustration of the device. (b) Image of an array of CMOS inverters with a magnified view in the inset. (c) A set of micrographs before and after applying external strains up to 30% in the x-direction. (d) Computed results for principal strain in regions of the FS structure that correspond to micrographs in (c). (e) Electrical characteristics of a representative CMOS inverter at a strain up to 20%. (f) Linear and log-scale transfer curves of a typical p-MOSFET (left) and current−voltage characteristics of a typical n-type MOSFET (right) with different gate biases.
The negative threshold voltage of the inverters arises from the high negative threshold voltage of the p-type MOSFETs. Figure 3f presents transfer curves in linear (red) and logarithmic (blue) scales for a typical p-channel MOSFET (left) and current−voltage characteristics for a typical n-channel MOSFET (right). The mobilities are $\sim 70 \text{ cm}^2/\text{V} \cdot \text{s}$ (p-type) and $\sim 400 \text{ cm}^2/\text{V} \cdot \text{s}$ (n-type); the on/off ($I_{on}/I_{off}$) ratios are $\sim 10^5$.

This type of stretchable mechanics is important for applications such as those in biomedicine, where conformal lamination onto the time-dynamic, curvilinear surfaces of the skin or other organs of the body is critically important. As an example of a type of transient biosensor with utility in this area, Figure 4a presents a schematic illustration of stretchable pH monitors built using doped Si NRs (length, width, and thickness: 500 μm, 40 μm, and 300 nm, respectively) functionalized with 3-aminopropyltriethoxysilane (APTES). Mg serves as the electrodes and interconnects, SiO$_2$ serves as the interlayer dielectrics and encapsulants, and POC serves as the substrate. The inset shows an image of the device. Figure 4b provides an optical micrograph of an individual pH monitor (top) and a magnified view (bottom). Figure 4c shows measured changes in conductance of devices constructed with phosphorus (left) and boron (right) doped Si NRs ($\sim 10^{20}/\text{cm}^3$ for both cases), as a function of time when exposed to solutions with different pH. Here, the $-\text{NH}_2$ and $-\text{SiOH}$ groups on the functionalized surfaces of the Si NRs undergo protonation to $-\text{NH}_3^+$ at low pH and deprotonation to $-\text{SiO}^-$ at high pH. Resulting changes in the surface charge electrostatically gate the transport in the silicon by depleting or accumulating charge carriers. The result is that the conductance of phosphorus and boron-doped Si NRs descends and ascends, respectively, in a series of distinct steps with pH (from 3 to 10) in aqueous solutions. The sensitivities of representative devices are 0.1 ± 0.01 and 0.3 ± 0.02 μS/pH for phosphorus and boron doped Si NRs, respectively. The latter results are in a range of those observed in boron-doped silicon nanowires (Si NWs) constructed in the usual way on silicon wafer substrates (0.1−4 μS/pH). This sensitivity depends on doping concentration, fabrication method (bottom-up, top-down), and geometry (surface-to-volume ratio) of the Si NWs, as well as local gate controls. To establish the time frame of viable measurements of pH with these transient devices Figure 4d shows changes in conductance of heavily boron-doped Si NRs ($\sim 200$ nm thick) while submerged in phosphate buffer solution (PBS, 0.1 M, pH 4−10, Sigma-Aldrich, U.S.A.) at 37 °C. (e) Images at various stages of dissolution of a device during immersion in PBS (pH 10) at room temperature.
The conductance of the Si NRs (heavily boron-doped) changes by $\sim 1\%$ or less during immersion for 5 days. This dissolution behavior is consistent with thickness changes reported previously for highly boron-doped Si NMs in similar solutions. The dissolution rates depend on the types and concentrations of ions in solution. For example, the rate in PBS (pH 7.4, 0.1 M) is $\sim 20$ times slower than in PBS (pH 7.4, 1 M) and $\sim 1.5$ times faster than that in PBS (pH 7.4, 0.05 M). Even for fixed concentration, the type of ions in the PBS can have strong effects. Recent changes in formulation of a commercial PBS (Sigma-Aldrich, U.S.A.) increased the dissolution rate by $\sim 20$ times, the details of which will be described elsewhere. The most relevant rates, of course, are those that occur in biofluids of interest. Dissolution studies of highly doped Si (boron and phosphorus) in bovine serum at 37 °C appear in Supporting Information Figure S6. In both cases, the dissolution rates are $\sim 2.5$ nm/day, which is somewhat slower than calculated values from previous reports likely due to differences in doping levels. Unencapsulated pH monitors have potential for use over limited periods of time at low or near neutral pH because the measured conductance (in the case of bovine serum) exhibits negligible changes (less than 1%) after a day in such conditions. At higher pH (pH 10), a linear decrease in conductance (less than 10%) during this period occurs, as expected. The time durations for stable operation can be extended by including overcoats of ultrathin dielectric insulators that exhibit slow dissolution rates, such as SiO$_2$ or SiN$_x$. Although this type of encapsulation can affect the response, previous studies suggest that ultrathin coatings of SiO$_2$ with surface treatments can enable useful levels of sensitivity. Figure 4e presents a set of images at various stages of dissolution of a pH sensor while immersed in PBS (pH 10, Sigma-Aldrich, U.S.A.) at room temperature. The Mg electrodes completely disappear after 12 h via reactive dissolution (i.e., hydrolysis), and other components (Si, SiO$_2$, POC) dissolve slowly in several weeks with rates similar to those reported in previous articles, in which the kinetics depend strongly on temperature, and pH and ionic content/concentration of solutions, as well as the morphologies of the materials. The rate of dissolution of POC can be controlled/tuned by adjusting the cross-linking density or using different monomers.

As another application example, Figure 5a,b presents optical images and schematic illustrations of a stretchable, transient EP sensor. The device includes thin layers of Mg (300 nm) and SiO$_2$ (100 nm) in the form of FS meshes (Figure 5a, inset) for measurement, ground and reference electrodes, and connecting leads. The Mg electrodes couple to the skin through a thin sheet of POC in a sensing mechanism that relies on displacement currents generated by capacitive coupling through the SiO$_2$. This arrangement enables EP measurements with levels of fidelity that are comparable to those of conventional

Figure 5. Capacitive biodegradable, stretchable EP sensors. (a) Optical image of a device and magnified view of the FS mesh electrode structure (inset). (b) Exploded view schematic illustration. (c) Photograph of a device mounted on the chest for measurement of ECG (left). ECG collected using a biodegradable (middle) and standard gel-based (right) device. (d) Photograph of the EP sensor located on the forearm for measurement of EMG (left). EMG collected using a biodegradable (middle) and standard gel-based (right) device, while clenching the first every 2 s. (e) A series of images of dissolution of a device in PBS (pH 10) at room temperature.
Fabrication of pH Sensors. The fabrication began with casting PMMA/D-PI layers onto a glass slide. Photolithographic patterning of a layer of Mg (300 nm) defined the layout of the EP sensors. A thin layer of SiO₂ (100 nm) formed by PECVD served as the interlayer dielectric with contact pads opened by etching with BOE to allow bonding to an anisotropic conductive film (ACF) cable. All other fabrication procedures were similar to those described for the CMOS inverters.

Electrical Characterization of pH Sensors. The Si NRs were cleaned by exposure to ultraviolet induced ozone for 3 min. Coating with APTES (Sigma-Aldrich, U.S.A.) occurred by immersion in a 1% ethanol solution of APTES for 20 min, rinsing with ethanol three times, and then heating at 80 °C for 10 min. ACF cables connected to bonding pads enabled external data acquisition. Measurements were performed during partial immersion in PBS with various pH values between 3 and 10. A Ag/AgCl reference electrode was placed in the center of the solutions and a floating gate voltage defined the quiescent conductance of the Si NRs. After a short period of stabilization, a semiconductor analyzer (415SC, Agilent, U.S.A.) recorded the conductance of the Si NRs for several seconds in the solutions with various pH from 3 to 10.

Electrical Measurements of Electrophysiological (EP) Sensors. A capacitive EP sensor was connected with ACF cable to a preamplifier with ultrahigh input impedance and extremely low-input capacitance (OPA124, Texas Instruments, U.S.A.; impedance = 10¹⁴Ω, 3 pF). This system provided a near-unity gain at the preamplifier, allowing EP signal measurement through an insulation layer (SiO₂). Details of the circuit appear elsewhere. The amplifier and filter units enabled tunable gain (60–80 dB) for ECGs and EMGs. The passband of the circuit was digitally varied using LabVIEW software (National Instruments, U.S.A.). The system was set to a gain of 60 dB in the range of 0.5–100 Hz for ECG; a gain of 80 dB in the range of 10–500 Hz for EMG. The acquired signals were sampled at 1 kHz by an analog-to-digital converter (NI USB-6363, National Instruments, U.S.A.) and displayed on a computer screen.

Dissolution Tests for Doped Si NRs. The test structures for studying dissolution behaviors of doped Si NRs were formed on a SOI wafer (top Si: ∼200 nm, SOITEC, France) by photolithography and RIE. Insoluble materials, such as gold (∼150 nm thick) and PI (∼1 μm thick) served as source/drain electrodes and encapsulation layers, respectively. The samples were immersed into ∼50 mL of phosphate buffer solutions (PBS, 0.1 M, pH 4–10, Sigma-Aldrich, U.S.A.) at a temperature of 37 °C. The samples were removed from the solution at regular time intervals to measure the conductance with a semiconductor analyzer (415SC, Agilent, U.S.A.). The solution was replaced every other day.
Notes
The authors declare no competing financial interest.

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REFERENCES